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         AUG 06
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         AUG 13
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     5
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                 FORIS renamed to SOFIS
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                 CA/CAplus enhanced with printed CA page images from
                 1967-1998
NEWS 13 SEP 17
                 CAplus coverage extended to include traditional medicine
                 patents
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                 CA/CAplus enhanced with pre-1907 records from Chemisches
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                 MEDLINE segment
         DEC 17 MEDLINE and LMEDLINE updated with 2008 MeSH vocabulary
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         DEC 17
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         DEC 17
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                 prophetic substances
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                 USPATFULL, USPAT2, and USPATOLD enhanced with new
         JAN 28
                 custom IPC display formats
NEWS 32
         JAN 28
                 MARPAT searching enhanced
NEWS 33
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                 of publication
NEWS 34 JAN 28
                 TOXCENTER enhanced with reloaded MEDLINE segment
NEWS 35
         JAN 28
                 MEDLINE and LMEDLINE reloaded with enhancements
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NEWS 36 FEB 08 STN Express, Version 8.3, now available

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AND CURRENT DISCOVER FILE IS DATED 24 JANUARY 2008

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FULL ESTIMATED COST

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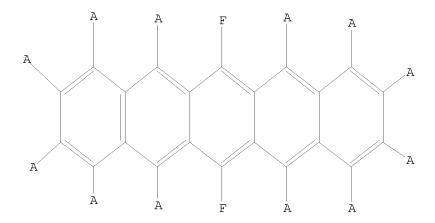
Uploading C:\Program Files\Stnexp\Queries\10578259-broadest.str

L1 STRUCTURE UPLOADED

=> d 11

L1 HAS NO ANSWERS

L1 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 11

SAMPLE SEARCH INITIATED 09:29:59 FILE 'REGISTRY' SAMPLE SCREEN SEARCH COMPLETED - 81 TO ITERATE

0 ANSWERS 100.0% PROCESSED 81 ITERATIONS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
BATCH **COMPLETE**

1081 TO 2159 PROJECTED ITERATIONS: PROJECTED ANSWERS: 0 TO

L2 0 SEA SSS SAM L1

=> s 11 full

FULL SEARCH INITIATED 09:30:03 FILE 'REGISTRY' FULL SCREEN SEARCH COMPLETED - 1577 TO ITERATE

100.0% PROCESSED 1577 ITERATIONS 11 ANSWERS

SEARCH TIME: 00.00.01

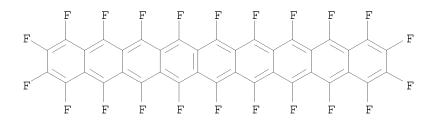
11 SEA SSS FUL L1

=> d 13 scan

REGISTRY COPYRIGHT 2008 ACS on STN T.3 11 ANSWERS

INDEX NAME NOT YET ASSIGNED IN

MF C38 F22



PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

HOW MANY MORE ANSWERS DO YOU WISH TO SCAN? (1):5

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

IN INDEX NAME NOT YET ASSIGNED

MF C30 F18

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

IN Pentacene, 1,2,3,4,5,6,7,8,9,11,12,13,14-tridecafluoro-10-iodo-

MF C22 F13 I

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

IN Pentacene, 2-chloro-1,3,4,5,6,7,8,9,10,11,12,13,14-tridecafluoro-

MF C22 Cl F13

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

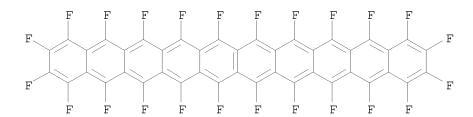
IN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro-MF C22 F14

PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

L3 11 ANSWERS REGISTRY COPYRIGHT 2008 ACS on STN

IN INDEX NAME NOT YET ASSIGNED

MF C42 F24



PROPERTY DATA AVAILABLE IN THE 'PROP' FORMAT

HOW MANY MORE ANSWERS DO YOU WISH TO SCAN? (1):0

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COST IN U.S. DOLLARS

COST IN U.S. DOLLARS SINCE FILE TOTAL ENTRY SESSION TULL ESTIMATED COST 178.82 179.03

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FILE LAST UPDATED: 19 Feb 2008 (20080219/ED)

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=> s 13

PUBLISHER:

L4 19 L3

=> s 13 not py > 2003

19 L3

5355707 PY > 2003

L5 1 L3 NOT PY > 2003

=> s 13 not py > 2004

19 L3

4248038 PY > 2004

L6 4 L3 NOT PY > 2004

=> d 16 ibib abs hitstr 1-

YOU HAVE REQUESTED DATA FROM 4 ANSWERS - CONTINUE? Y/(N):y

L6 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:577543 CAPLUS

DOCUMENT NUMBER: 141:260225

TITLE: The essential role of H-F substitution in the

electron-phonon interactions and electron transfer in

the negatively charged acenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Fukui Institute for Fundamental Chemistry, Kyoto

University, Sakyo-ku, Kyoto, 606-8103, Japan

SOURCE: Journal of Chemical Physics (2004), 121(5), 2356-2366

CODEN: JCPSA6; ISSN: 0021-9606 American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE: English

The single charge transfer through acenes, partially H-F substituted acenes, and fluoroacenes is discussed. The reorganization energies between the neutral mols. and the corresponding monoanions for partially H-F substituted acenes lie between those for acenes and fluoroacenes. The delocalization of the lowest unoccupied MOs (LUMO) by substituting hydrogen atoms by fluorine atoms with the highest electronegativity in every element is the main reason why the reorganization energy between the neutral mol. and the monoanion for partially H-F substituted acenes lies between those for acenes and fluoroacenes. This result implies that the neg. charged partially H-F substituted acenes would be better conductors with rapid electron transfer than the neg. charged fluoroacenes if we assume that the overlap of the LUMO between partially H-F substituted acenes is not significantly different from that between two neighboring fluoroacenes. The structures of the monoanions of acenes, fluoroacenes, and partially H-F substituted acenes are optimized under D2h geometry, and the Jahn-Teller effects in the monoanions of benzene and fluorobenzene are discussed. The vibration effect onto the charge transfer problem is also discussed. The C-C stretching modes around $15\bar{0}0$ cm-1 are the main modes converting the neutral mols. to the monoanions in acenes, fluoroacenes, and partially H-F substituted acenes. It can be confirmed from the calculational results that the C-C stretching modes around 1500 cm-1 the most strongly couple to the LUMO in these mols. The main reason why the total electron-phonon coupling consts. (1LUMO) for the monoanions of acenes in which four outer hydrogen atoms are substituted by fluorine atoms are larger than those for the monoanions of acenes in which several inner hydrogen atoms are substituted by fluorine atoms is suggested. The

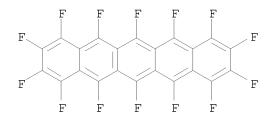
relationships between the electron transfer and the electron-phonon interactions are discussed. The plot of the reorganization energies against the lLUMO values is found to be nearly linear. In view of these results, the relationships between the normal and superconducting states are briefly discussed.

IT 646533-88-2 690975-12-3 RL: PRP (Properties)

(essential role of H-F substitution in electron-phonon interactions and electron transfer in neq. charged acenes)

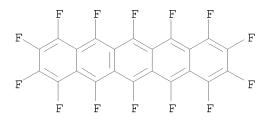
RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX



RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT: 99 THERE ARE 99 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:480070 CAPLUS

DOCUMENT NUMBER: 141:197895

TITLE: Perfluoropentacene: High-Performance p-n Junctions and

Complementary Circuits with Pentacene

AUTHOR(S): Sakamoto, Youichi; Suzuki, Toshiyasu; Kobayashi,

Masafumi; Gao, Yuan; Fukai, Yasushi; Inoue, Youji;

Sato, Fumio; Tokito, Shizuo

CORPORATE SOURCE: Institute for Molecular Science, Okazaki, 444-8787,

Japan

SOURCE: Journal of the American Chemical Society (2004),

126(26), 8138-8140

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

AB The authors report the synthesis and characterization of perfluoropentacene as an n-type semiconductor for organic field-effect transistors (OFETs). Perfluoropentacene is a planar and crystalline material that adopts a herringbone structure as observed for pentacene. OFETs with perfluoropentacene were constructed using top-contact geometry, and an electron mobility of 0.11 cm2 V-1 s-1 was observed Bipolar OFETs with

perfluoropentacene and pentacene function at both neg. and pos. gate voltages. The improved p-n junctions are probably due to the similar d-spacings of both acenes. Complementary inverter circuits were fabricated, and the transfer characteristics exhibit a sharp inversion of the output signal with a high-voltage gain.

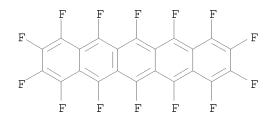
IT 646533-88-2P, Perfluoropentacene

RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(perfluoropentacene: high-performance p-n junctions and complementary circuits with pentacene)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:297458 CAPLUS

DOCUMENT NUMBER: 140:429323

TITLE: Inverse isotope effects and electron-phonon coupling

in the positively charged deutero- and fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Fukui Institute for Fundamental Chemistry, Kyoto

University, Sakyo-ku, Kyoto, 606-8103, Japan

SOURCE: Journal of Chemical Physics (2004), 120(16), 7659-7672

CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE: English

Electron-phonon interactions in the monocations of deutero- and fluoroacenes are studied and compared with those in the monocations of acenes and those in the monoanions of fluoroacenes. Because of the significant phase pattern difference between the highest occupied MOs (HOMO) and the lowest unoccupied MOs (LUMO), the frequency modes lower than 500 cm-1 and the high-frequency modes around 1400 cm-1 couple more strongly to the LUMO than to the HOMO, while the frequency modes around 500 cm-1 and the frequency modes around 1600 cm-1 couple more strongly to the HOMO than to the LUMO in fluoroacenes with D2h geometry. The total electron-phonon coupling consts. for the monocations (1HOMO) are estimated and compared with those for the monoanions (1LUMO) in deutero- and fluoroacenes. The 1HOMO values are estimated to be 0.418, 0.399, 0.301, 0.255, and 0.222 eV for C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f), resp. The lHOMO values are smaller than the lLUMO values in small fluoroacenes. But the lHOMO value decreases with an increase in mol. size less rapidly than the lLUMO value in fluoroacenes, and the lHOMO value of $0.074~{\rm eV}$ is much larger than the 1LUMO value of $0.009~{\rm eV}$ in polyfluoroacene. The logarithmically averaged phonon frequencies for the monocations (ω ln,HOMO) are estimated to be larger than those for the monoanions $(\omega ln, LUMO)$ in fluoroacenes. This is because the C-C stretching modes around 1600 cm-1 couple most strongly to the HOMO, and those around $1400\ \mathrm{cm}{-1}$ couple the most strongly to the LUMO in

fluoroacenes. The significant phase pattern difference between the HOMO and the LUMO is the main reason for the calculational results. The lHOMO values increase much more significantly by H-F substitution than by H-D substitution in acenes. The possible inverse isotope effects in the electron-phonon interactions as a consequence of deuteration in the monocations of nanosized mols. are suggested.

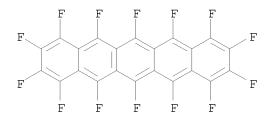
IT 690975-12-3

RL: PRP (Properties)

(inverse isotope effects and electron-phonon coupling in pos. charged deutero- and fluoroacenes)

RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT: 38 THERE ARE 38 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L6 ANSWER 4 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2003:898555 CAPLUS

DOCUMENT NUMBER: 140:118999

TITLE: Electron-phonon interactions in the monoanions of

fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Graduate School of Engineering, Department of Material

Chemistry, Kyoto University, Sakyo-ku, Kyoto,

606-8501, Japan

SOURCE: Journal of Chemical Physics (2003), 119(21),

11318-11328

CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE: English

AΒ Electron-phonon interactions in the monoanions of fluoroacenes such as C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f) are studied, and compared with those in the monoanions of acenes and deutero-acenes. The C-C stretching modes around 1500 cm-1 the most strongly couple to the lowest unoccupied MOs (LUMO) in fluoroacenes. The estimated total electron-phonon coupling consts. (1LUMO) are 0.475, 0.473, 0.350, 0.273, and 0.215 eV for 1f, 2f, 3f, 4f, and 5f, resp. The lLUMO values for fluoroacenes are much larger than those for acenes and deutero-acenes. Possible superconducting transition temps. (Tcs) for the monoanions of deutero-acenes and fluoroacenes are larger than those for the monoanions of acenes. The transition temperature (Tc) value increases much more significantly by H-F substitution than by H-D substitution in acenes. The lLUMO and Tc values significantly decrease with an increase in mol. size from the monoanions of 1f to 5f. The logarithmically averaged phonon frequencies (ω ln) do not significantly change with an increase in mol. size in the monoanions of fluoroacenes. The larger displacements of C atoms in the vibronic active modes in fluoroacenes than those in deutero-acenes due to larger atomic mass of F than that of D, and the unchanged properties of the orbital patterns of the LUMO as a consequence of H-F and H-D substitution in acenes, are the main reasons why the 1LUMOvalue increases much more significantly by H-F substitution than by H-D

substitution, and the reason why the Tc value increases much more significantly by H-F substitution than by H-D substitution in acenes. The detailed properties of vibronic active modes and the electronic structures in the LUMO as well as the mol. wts. are closely related to the lLUMO, ω ln, and Tc values in the monoanions of fluoroacenes, deuteroacenes, and acenes.

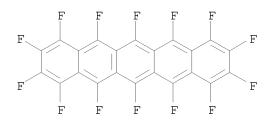
IT 646533-88-2

RL: PRP (Properties)

(carbon-carbon bond lengths in)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



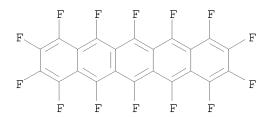
IT 645401-34-9

RL: PRP (Properties)

(electron-phonon interactions in)

RN 645401-34-9 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1-) (9CI) (CA INDEX NAME)



REFERENCE COUNT: 90 THERE ARE 90 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

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http://www.cas.org/support/stngen/stndoc/properties.html

=>

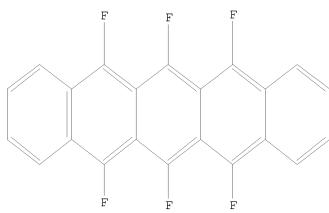
Uploading C:\Program Files\Stnexp\Queries\10578259-claim-2.str

L7 STRUCTURE UPLOADED

=> d 17

L7 HAS NO ANSWERS

L7 STR



Structure attributes must be viewed using STN Express query preparation.

=> s 17

SAMPLE SEARCH INITIATED 09:46:54 FILE 'REGISTRY'
SAMPLE SCREEN SEARCH COMPLETED - 60 TO ITERATE

100.0% PROCESSED 60 ITERATIONS 0 ANSWERS

SEARCH TIME: 00.00.01

FULL FILE PROJECTIONS: ONLINE **COMPLETE**
BATCH **COMPLETE**

BAICH ""COMPLETE""

PROJECTED ITERATIONS: 736 TO 1664
PROJECTED ANSWERS: 0 TO 0

L8 0 SEA SSS SAM L7

=> s 17 full

FULL SEARCH INITIATED 09:46:59 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 1124 TO ITERATE

100.0% PROCESSED 1124 ITERATIONS

11 ANSWERS

SEARCH TIME: 00.00.01

L9 11 SEA SSS FUL L7

=> file caplus

COST IN U.S. DOLLARS
SINCE FILE TOTAL
ENTRY SESSION
FULL ESTIMATED COST
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395.91

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=> s 19 not 14 19 L9

L10 0 L9 NOT L4

=> s 19 not py > 2005

19 L9

3044168 PY > 2005

L11 7 L9 NOT PY > 2005

=> s 19 not py > 2004

19 L9

4248038 PY > 2004

L12 4 L9 NOT PY > 2004

=> d 112 ibib abs hitstr 1-

YOU HAVE REQUESTED DATA FROM 4 ANSWERS - CONTINUE? Y/(N):y

L12 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:577543 CAPLUS

DOCUMENT NUMBER: 141:260225

TITLE: The essential role of H-F substitution in the

electron-phonon interactions and electron transfer in

the negatively charged acenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Fukui Institute for Fundamental Chemistry, Kyoto

University, Sakyo-ku, Kyoto, 606-8103, Japan

SOURCE: Journal of Chemical Physics (2004), 121(5), 2356-2366

CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE: English

The single charge transfer through acenes, partially H-F substituted AB acenes, and fluoroacenes is discussed. The reorganization energies between the neutral mols. and the corresponding monoanions for partially H-F substituted acenes lie between those for acenes and fluoroacenes. The delocalization of the lowest unoccupied MOs (LUMO) by substituting hydrogen atoms by fluorine atoms with the highest electronegativity in every element is the main reason why the reorganization energy between the neutral mol. and the monoanion for partially H-F substituted acenes lies between those for acenes and fluoroacenes. This result implies that the neg. charged partially H-F substituted acenes would be better conductors with rapid electron transfer than the neg. charged fluoroacenes if we assume that the overlap of the LUMO between partially H-F substituted acenes is not significantly different from that between two neighboring fluoroacenes. The structures of the monoanions of acenes, fluoroacenes, and partially H-F substituted acenes are optimized under D2h geometry, and the Jahn-Teller effects in the monoanions of benzene and fluorobenzene are discussed. The vibration effect onto the charge transfer problem is also discussed. The C-C stretching modes around 1500 cm-1 are the main modes converting the neutral mols. to the monoanions in acenes, fluoroacenes, and partially H-F substituted acenes. It can be confirmed from the calculational results that the C-C stretching modes around 1500 cm-1 the most strongly couple to the LUMO in these mols. The main reason why the total electron-phonon coupling consts. (lLUMO) for the monoanions of acenes in which four outer hydrogen atoms are substituted by fluorine atoms are larger than those for the monoanions of acenes in which several inner hydrogen atoms are substituted by fluorine atoms is suggested. relationships between the electron transfer and the electron-phonon interactions are discussed. The plot of the reorganization energies against the lLUMO values is found to be nearly linear. In view of these results, the relationships between the normal and superconducting states are briefly discussed.

IT 646533-88-2 690975-12-3

RL: PRP (Properties)

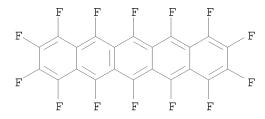
(essential role of H-F substitution in electron-phonon interactions and electron transfer in neg. charged acenes)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)

RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT: 99 THERE ARE 99 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L12 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:480070 CAPLUS

DOCUMENT NUMBER: 141:197895

TITLE: Perfluoropentacene: High-Performance p-n Junctions and

Complementary Circuits with Pentacene

AUTHOR(S): Sakamoto, Youichi; Suzuki, Toshiyasu; Kobayashi,

Masafumi; Gao, Yuan; Fukai, Yasushi; Inoue, Youji;

Sato, Fumio; Tokito, Shizuo

CORPORATE SOURCE: Institute for Molecular Science, Okazaki, 444-8787,

Japan

SOURCE: Journal of the American Chemical Society (2004),

126(26), 8138-8140

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

AB The authors report the synthesis and characterization of perfluoropentacene as an n-type semiconductor for organic field-effect transistors (OFETs). Perfluoropentacene is a planar and crystalline material that adopts a herringbone structure as observed for pentacene. OFETs with perfluoropentacene were constructed using top-contact geometry, and an electron mobility of 0.11 cm2 V-1 s-1 was observed Bipolar OFETs with perfluoropentacene and pentacene function at both neg. and pos. gate voltages. The improved p-n junctions are probably due to the similar d-spacings of both acenes. Complementary inverter circuits were fabricated, and the transfer characteristics exhibit a sharp inversion of the output signal with a high-voltage gain.

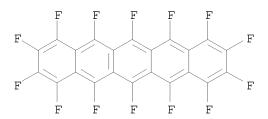
IT 646533-88-2P, Perfluoropentacene

RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(perfluoropentacene: high-performance p-n junctions and complementary circuits with pentacene)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)



REFERENCE COUNT: 24 THERE ARE 24 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L12 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2008 ACS on STN

ACCESSION NUMBER: 2004:297458 CAPLUS

DOCUMENT NUMBER: 140:429323

TITLE: Inverse isotope effects and electron-phonon coupling

in the positively charged deutero- and fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Fukui Institute for Fundamental Chemistry, Kyoto

University, Sakyo-ku, Kyoto, 606-8103, Japan

SOURCE: Journal of Chemical Physics (2004), 120(16), 7659-7672

CODEN: JCPSA6; ISSN: 0021-9606

PUBLISHER: American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE: English

Electron-phonon interactions in the monocations of deutero- and fluoroacenes are studied and compared with those in the monocations of acenes and those in the monoanions of fluoroacenes. Because of the significant phase pattern difference between the highest occupied MOs (HOMO) and the lowest unoccupied MOs (LUMO), the frequency modes lower than 500 cm-1 and the high-frequency modes around 1400 cm-1 couple more strongly to the LUMO than to the HOMO, while the frequency modes around 500 cm-1 and the frequency modes around 1600 cm-1 couple more strongly to the HOMO than to the LUMO in fluoroacenes with D2h geometry. The total electron-phonon coupling consts. for the monocations (1HOMO) are estimated and compared with those for the monoanions (1LUMO) in deutero- and fluoroacenes. The 1HOMO values are estimated to be 0.418, 0.399, 0.301, 0.255, and 0.222 eV for C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f), resp. The lHOMO values are smaller than the lLUMO values in small fluoroacenes. But the lHOMO value decreases with an increase in mol. size less rapidly than the lLUMO value in fluoroacenes, and the lHOMO value of $0.074~{\rm eV}$ is much larger than the lLUMO value of $0.009~{\rm eV}$ in polyfluoroacene. The logarithmically averaged phonon frequencies for the monocations (ω ln, HOMO) are estimated to be larger than those for the monoanions (ω ln,LUMO) in fluoroacenes. This is because the C-C stretching modes around 1600 cm-1 couple most strongly to the HOMO, and those around 1400 cm-1 couple the most strongly to the LUMO in fluoroacenes. The significant phase pattern difference between the HOMO and the LUMO is the main reason for the calculational results. The 1HOMO values increase much more significantly by H-F substitution than by H-D substitution in acenes. The possible inverse isotope effects in the electron-phonon interactions as a consequence of deuteration in the monocations of nanosized mols. are suggested.

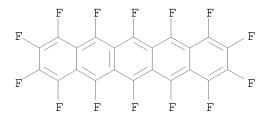
IT 690975-12-3

RL: PRP (Properties)

(inverse isotope effects and electron-phonon coupling in pos. charged deutero- and fluoroacenes)

RN 690975-12-3 CAPLUS

CN Pentacene, tetradecafluoro-, radical ion(1+) (9CI) (CA INDEX NAME)



REFERENCE COUNT:

THERE ARE 38 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

38

ACCESSION NUMBER: 2003:898555 CAPLUS

DOCUMENT NUMBER: 140:118999

TITLE: Electron-phonon interactions in the monoanions of

fluoroacenes

AUTHOR(S): Kato, Takashi; Yamabe, Tokio

CORPORATE SOURCE: Graduate School of Engineering, Department of Material

Chemistry, Kyoto University, Sakyo-ku, Kyoto,

606-8501, Japan

SOURCE: Journal of Chemical Physics (2003), 119(21),

11318-11328

CODEN: JCPSA6; ISSN: 0021-9606 American Institute of Physics

DOCUMENT TYPE: Journal LANGUAGE: English

PUBLISHER:

Electron-phonon interactions in the monoanions of fluoroacenes such as C6F6 (1f), C10F8 (2f), C14F10 (3f), C18F12 (4f), and C22F14 (5f) are studied, and compared with those in the monoanions of acenes and deutero-acenes. The C-C stretching modes around 1500 cm-1 the most strongly couple to the lowest unoccupied MOs (LUMO) in fluoroacenes. estimated total electron-phonon coupling consts. (1LUMO) are 0.475, 0.473, 0.350, 0.273, and 0.215 eV for 1f, 2f, 3f, 4f, and 5f, resp. The lLUMO values for fluoroacenes are much larger than those for acenes and deutero-acenes. Possible superconducting transition temps. (Tcs) for the monoanions of deutero-acenes and fluoroacenes are larger than those for the monoanions of acenes. The transition temperature (Tc) value increases much more significantly by H-F substitution than by H-D substitution in acenes. The lLUMO and Tc values significantly decrease with an increase in mol. size from the monoanions of 1f to 5f. The logarithmically averaged phonon frequencies (ω ln) do not significantly change with an increase in mol. size in the monoanions of fluoroacenes. The larger displacements of C atoms in the vibronic active modes in fluoroacenes than those in deutero-acenes due to larger atomic mass of F than that of D, and the unchanged properties of the orbital patterns of the LUMO as a consequence of H-F and H-D substitution in acenes, are the main reasons why the lLUMO value increases much more significantly by H-F substitution than by H-D substitution, and the reason why the Tc value increases much more significantly by H-F substitution than by H-D substitution in acenes. The detailed properties of vibronic active modes and the electronic structures in the LUMO as well as the mol. wts. are closely related to the lLUMO, ω ln, and Tc values in the monoanions of fluoroacenes, deuteroacenes, and acenes.

IT 646533-88-2

RL: PRP (Properties)

(carbon-carbon bond lengths in)

RN 646533-88-2 CAPLUS

CN Pentacene, 1,2,3,4,5,6,7,8,9,10,11,12,13,14-tetradecafluoro- (CA INDEX NAME)

IT 645401-34-9

RL: PRP (Properties)

(electron-phonon interactions in)

RN 645401-34-9 CAPLUS

90

REFERENCE COUNT:

THERE ARE 90 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT